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approximately $5.04 \times$
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$6.67 \times 10^{-11} \frac{\text{N m}^2}{\text{kg}^2}$
 $(6.0 \times 10^{24} \text{ kg})$

$(6.37 \times 10^6 \text{ m})$
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$\approx 4.28 \times 10^3$ The escape
velocity for Earth is

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the escape velocity will
be:

$$v_e = \sqrt{2 \times 9.8 \times 63,781,00}$$

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 $V = \sqrt{2GM/r}$
 11.23×10^3 m/s
 2.44×10^3 m/s
 6.67×10^{-11} (N m² kg⁻²)
 (3.35×10^24) kg
 6.37×10^6 m
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The temperature, at which the root mean square velocity of hydrogen molecules equals their escape velocity from the earth is closest to:

[Boltzmann's Constant
 $1.38 \times 10^{-23} \text{ J/K}$
Avogadro number 6.02×10^{23}]

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